

(12) INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(19) World Intellectual Property Organization
International Bureau



(43) International Publication Date
26 January 2006 (26.01.2006)

PCT

(10) International Publication Number
WO 2006/008749 A1

(51) International Patent Classification: **C07D 501/18**

(21) International Application Number:
PCT/IN2004/000209

(22) International Filing Date: 16 July 2004 (16.07.2004)

(25) Filing Language: English

(26) Publication Language: English

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(81) Designated States (*unless otherwise indicated, for every kind of national protection available*): AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NI, NO, NZ, OM, PG,

PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW.

(84) Designated States (*unless otherwise indicated, for every kind of regional protection available*): ARIPO (BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW), Eurasian (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European (AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, NL, PL, PT, RO, SE, SI, SK, TR), OAPI (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG).

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- *of inventorship (Rule 4.17(iv)) for US only*

Published:

- *with international search report*

For two-letter codes and other abbreviations refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.

(54) Title: PROCESS FOR PREPARING PURE CEPHALOSPORINE INTERMEDIATES

(57) Abstract: The present invention relates to a process for preparing key intermediates for cephalosporin antibiotics substantially free of undesired delta A² isomer. Thus, 7-aminocephalosporanic acid (7-ACA) is silylated with hexamethyl-disilazane in cyclohexane at reflux temperature [6R,7R]-3-[(Acetoxy)methyl]-7-(trimethylsilyl)aminoceph-3-em-4-oic acid obtained is reacted with the mixture of N-methylpyrrolidine and trimethylsilyliodide in cyclohexane, desilylated with isopropyl alcohol and treated with hydrochloric acid to obtain [6R-(6a,7b)]-1-[[7-Amino-2-carboxy-8-oxo-5-thia-1-azabicyclo[4.2.0]oct-2-en-3-yl]methyl]-1-methylpyrrolidinium inner salt hydrochloride [6R-(6a,7b)]-1-[[7-Amino-2-carboxy-8-oxo-5-thia-1-azabicyclo[4.2.0]oct-2-en-3-yl]methyl]-1-methylpyrrolidinium inner salt hydrochloride is N-acylated with syn-2-(2-aminothiazol-4-yl)-2-methoxyimino acetic acid 2-benzothiazolyl thioester (MAEM) followed by treatment with hydrochloric acid to give cefepime dihydrochloride monohydrate

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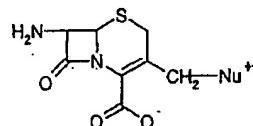
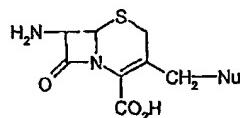
PROCESS FOR PREPARING PURE CEPHALOSPORINE INTERMEDIATESFIELD OF THE INVENTION

The present invention relates to a process for preparing key intermediates for cephalosporin antibiotics substantially free of undesired Δ^2 isomer. According to the novel process, no chromatographic separations are required for isolating Δ^2 isomer thereby increasing the productivity. Moreover the novel process avoids the use of expensive, environmentally hazardous fluorochlorocarbons such as freon. Thus, the novel process is environmentally safe, less expensive and commercially viable.

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BACKGROUND OF THE INVENTION

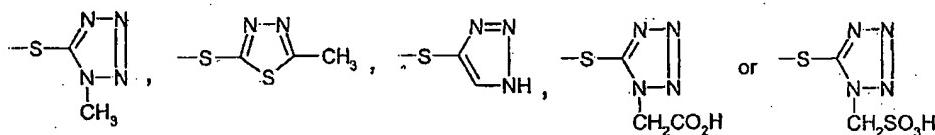
U.S. Patent No. 4,868, 294 described crystalline temperature stable hydrochloride or hydroiodide salt of a compound of formulas:



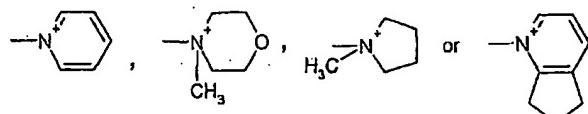
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which is substantially free of the corresponding Δ^2 isomer,
wherein

Nu is



and

Nu⁺ is

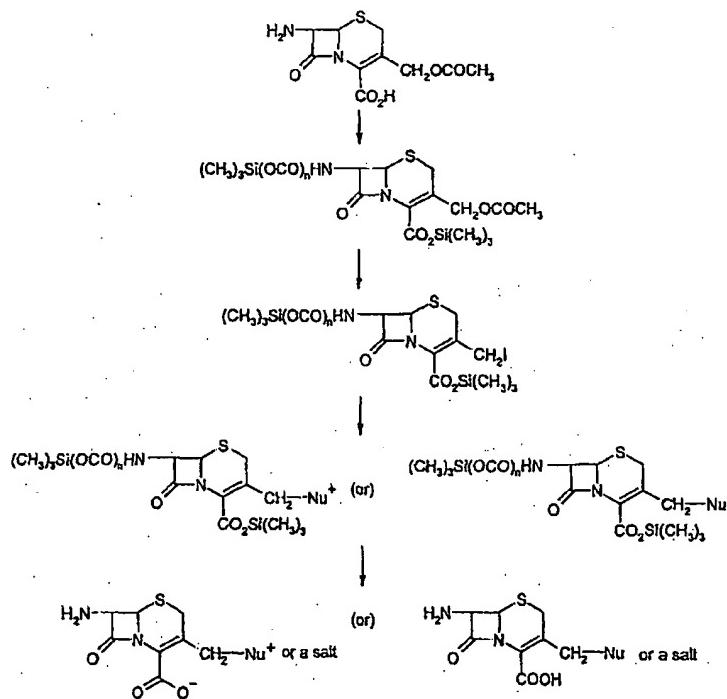
These compounds are key intermediates for the conversion by acylation into broad spectrum cephalosporin antibiotics which are substantially free of Δ^2 isomer.

Various cephalosporin antibiotics were disclosed in many patents, some of which are U. S. Patent No. 4,406,899, U. S. Patent No. 4,168309, U. S. Patent No. 5 4,223,135, U. S. Patent No. 4,336,253 and U. S. Patent No. 4,379,787.

J. Organic Chemistry 1988, 53, 983-991 described the effect of halogenated solvents, acetonitrile and toluene on the formation of Δ^2 isomer during the preparation of the key intermediates such as those mentioned above.

U.S. Patent No. 4,910,301 disclosed temperature stable crystalline salts of 7-[α -(2-aminothiazol-4-yl)- α -(Z)-methoxyiminoacetamido]-3-[(1-methyl-1-pyrro lidinio)methyl]-10 3-cephem-4-carboxylate (cefepime). These salts include among others cefepime dihydrochloride monohydrate and cefepime sulfuric acid salt.

According to U.S. Patent No. 4,868,294, the key intermediates substantially free of Δ^2 isomer mentioned above can be prepared by carrying out the reactions according 15 to the following reaction scheme in freon (1,1,2-trichlorotrifluoroethane) solvent medium:



It is known that freon is environmentally hazardous chlorofluoro carbon and is expensive.

U.S. Patent No. 5,441,874 and EP patent No. 0162395 described processes for 20 preparing some cephalosporin antibiotics.

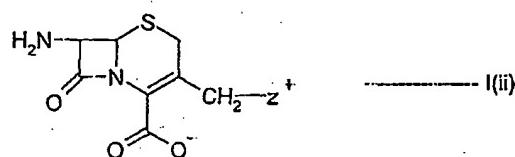
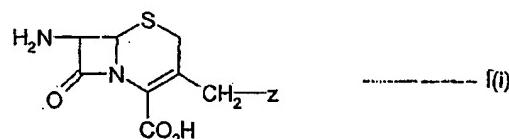
U.S. Patent No. 5,594,130 described preparation of cefepime using syn-isomer of 2-(2-aminothiazol-4-yl)-2-methoxyimino acetyl chloride hydrochloride.

U.S. Patent No. 4,680,389 described stable crystalline di (1-methyl-2-pyrrolidinone) and di (N-formyl pyrrolidine) adducts of cephalosporin derivatives such as 5 cefepime.

We have found that the formation of undesired Δ^2 isomer in the preparation of key intermediates for cephalosporin antibiotics can be reduced or avoided with the use of cyclohexane as solvent. According to the novel process, no chromatographic separations are required for isolating Δ^2 isomer thereby increasing the productivity. 10 Moreover the novel process avoids the use of expensive, environmentally hazardous fluorochlorocarbons such as freon. Thus, the novel process is environmentally safe, less expensive and commercially viable.

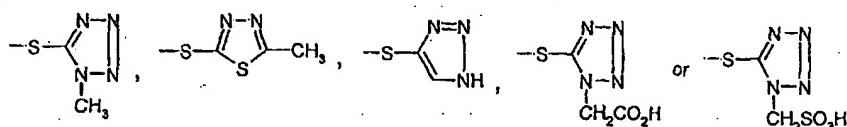
DETAILED DESCRIPTION OF THE INVENTION

15 According to the present invention, a process is provided for preparing the compounds of formulas- I(i) & I(ii) substantially free of the corresponding Δ^2 isomer; or salts thereof.



wherein

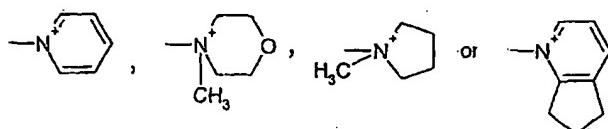
Z is



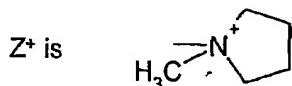
and

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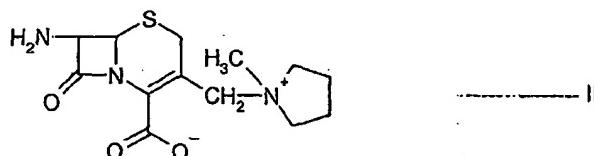
Z^+ is



The preferred compound prepared according to the present invention is the compound of formula I(ii), wherein



and is represented by the formula II:

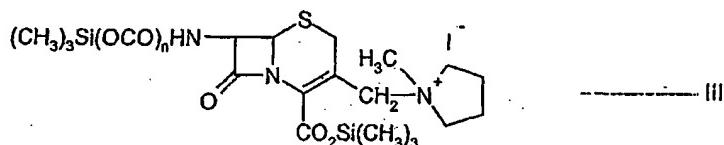


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and a salt thereof.

Preferable salts are hydrochloride and hydroiodide salts.

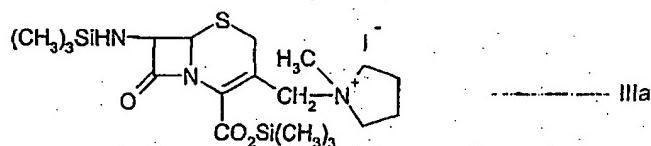
The compound of formula II may be prepared by treating a solution of the compound of the formula III:



10

wherein $n = 0$ or 1 ,

in cyclohexane with a $C_1 - C_4$ -alkanol or water to remove silyl protecting groups. The compounds of formula II are preferably converted into a salt. The compound of formula III, wherein $n = 0$ is the preferred compound and is represented by formula IIIa:



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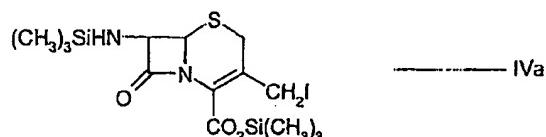
The reaction is carried out at a temperature of from about -10°C to about 45°C , preferably at a temperature of from about 0°C to about 25°C , and more preferably at a temperature of from about 0°C to about 10°C . Preferable alcohols are isopropyl alcohol, methanol and ethanol, more preferable being isopropyl alcohol. From about 1 to about 5 equivalents of $C_1 - C_4$ -alkanol are used per equivalent of compound III.

The compounds of the formula III may be prepared by reacting a solution of the compounds of the formula IV:



wherein $n = 0$ or 1 .

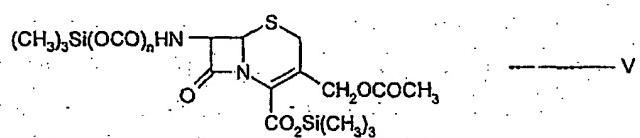
in a cyclohexane with N-methyl pyrrolidine. The compound of formula IV, wherein $n = 0$,
5 is the preferred compound and is represented by formula IVa:



It has been surprisingly found that when cyclohexane is used as solvent, compound III obtained is substantially free of the Δ^2 isomer. It is known from U. S. Patent No. 4,868,294 that when the solvents such as methylene dichloride, carbon tetrachloride, chloroform or dioxane are used, the product obtained contains large amounts of the undesired Δ^2 isomer.

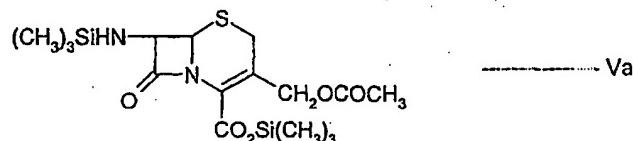
The reaction is carried out at a temperature of from about -10°C to about 45°C and preferably at a temperature of from about 0°C to about 25°C. The amount of N-methyl pyrrolidine is not critical, but preferably about 1 to about 2 equivalents of N-methyl pyrrolidine per equivalent of compound of formula IV

The compound of the formula IV may be prepared by reaction of a solution of the compound of the formula V:



wherein $n = 0$ or 1 ,

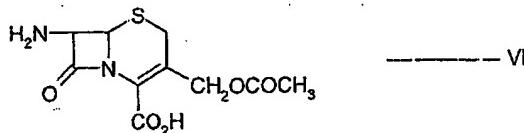
20 in a cyclohexane with trimethylsilyl iodide (TMSI). The compound of formula V, wherein n = 0 is the preferred compound and is represented by formula Va:



When cyclohexane is used as solvent, the compound of formula IV obtained is substantially free of the Δ^2 isomer. As it is known from the description in U.S. Patent No. 4,868,294, solvents such as 1,2-dichloroethane, chlorobenzene, dioxane and carbontetrachloride, yield compound IV containing significant amounts of the undesirable Δ^2 isomer.

The reaction is carried out at a temperature of from about 0°C to about 45°C, preferably at a temperature from about 5°C to about 40°C and more preferably at a temperature from about 5°C to about 25°C. At least one equivalent of trimethylsilyl iodide is required to convert all the compound V to IV, preferable amount being about 0.9 to about 2.5 equivalents per equivalent of compound V, more preferable amount being about 1.0 to about 2.0 equivalents of trimethylsilyl iodide.

The compounds of formula Va may be prepared by reacting 7-aminocephalosporanic acid (7-ACA) of the formula VI:



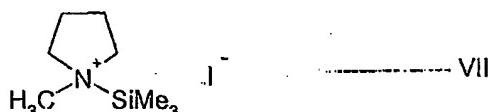
with hexamethyldisilazane (HMDS) at a temperature from about 0°C to the boiling temperature of the cyclohexane. The reaction is preferably carried out in the presence of catalytic amount (about 0.05 to about 0.1 equivalent each per equivalent of 7-ACA) of imidazole and acetamide; or in the presence of catalytic amount (about 0.01 to about 0.1 equivalent per equivalent of 7-ACA) of trimethylsilyl iodide. The reaction is preferably carried out at a temperature from about 25°C to the boiling temperature of cyclohexane, more preferably from about 35°C to the boiling temperature of cyclohexane and most preferably at the boiling temperature of cyclohexane. It has been found that silylation occurs to a larger extent at a faster rate when the silylation is carried out at the boiling temperature of cyclohexane than when the silylation is carried out at a lower temperature. The HMDS may be used in an amount from about 0.9 to about 1.5 equivalents per equivalent of 7-ACA, preferably from about 1.0 to 1.4 equivalents of HMDS per equivalent of 7-ACA. The catalytic amounts of acetamide and imidazole may preferably be used in the silylation step.

The compound of formula V, wherein n = 1 may be prepared by bubbling carbon dioxide gas into a solution of compound Va in cyclohexane.

In an alternative preparation of compound of formula III, a solution of compound of formula V in cyclohexane is treated with N-methyl pyrrolidine followed by the addition of at least one equivalent of trimethylsilyl iodide. The reaction can be conducted at a

temperature of from about 0°C to about 45°C and preferably from about 0°C to about 25°C. The N-methyl pyrrolidine may be used in an amount of from about 1.0 to about 2.0 equivalents per equivalent of compound V. The trimethylsilyl iodide may be used in an amount of from about 0.9 to about 2.5 equivalents per equivalent of compound V, and preferably from about 1.0 to 1.8 equivalents.

In an another alternative preparation of compound III, a solution of compound V in cyclohexane is reacted with N-methyl-N-trimethylsilyl pyrrolidino iodide having the formula VII:



at a temperature from about 0°C to about 45°C and preferably from about 0°C to about 25°C. The reaction may preferably be carried out in the presence of trimethylsilyl iodide in an amount from about 0.2 to about 0.8 equivalents per equivalent of compound V. The compound of formula VII may be used in an amount from about 1.0 to about 2.5 equivalents per equivalent of compound V and preferably from about 1.0 to about 2.0 equivalents of compound VII per equivalent of compound V.

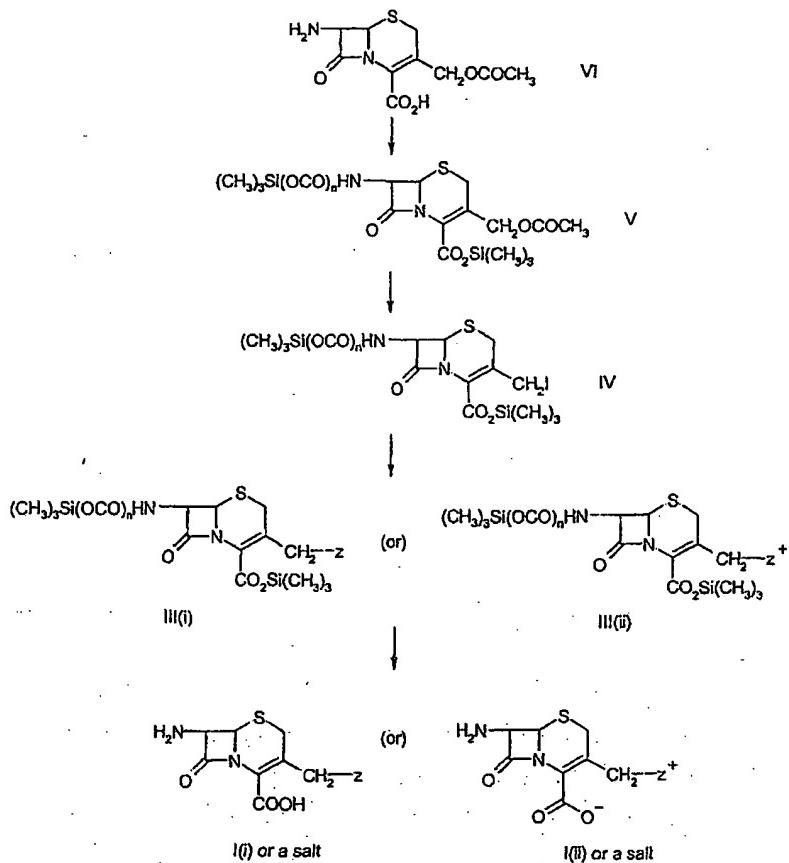
The compound of formula VII may be prepared by reacting N-methyl pyrrolidine with about an equimolar amount of trimethylsilyl iodide in cyclohexane at a temperature of from about -10°C to about 45°C. Preferably the reaction is carried out at a temperature of from about 0°C to about 25°C, more preferably from about 0°C to about 10°C.

In a preferred reaction scheme, the compound of formula II or the salt thereof is prepared from 7-ACA in a "one pot" reaction i.e., without the isolation of any intermediates using cyclohexane as main solvent thought out the reaction sequence.

The other compounds of formula I or their salts may be prepared by similar procedure described for the compound II and its salts.

The "compound substantially free of Δ^2 isomer" refers to the compound containing the content of Δ^2 isomer in less than about 10% of the compound plus the isomer, preferably less than about 3% and more preferably less than about 0.4%.

The compounds of the formula I can be prepared by the sequence shown below in reaction scheme I.

Reaction Scheme 1

In the compounds of the formulas III(i), III(ii), IV and V, n = 0 or 1 and in the compound of the formulas III(i) and III(ii), Z and Z⁺ have the same meaning as defined in formula in formulas I(i) and I(ii).

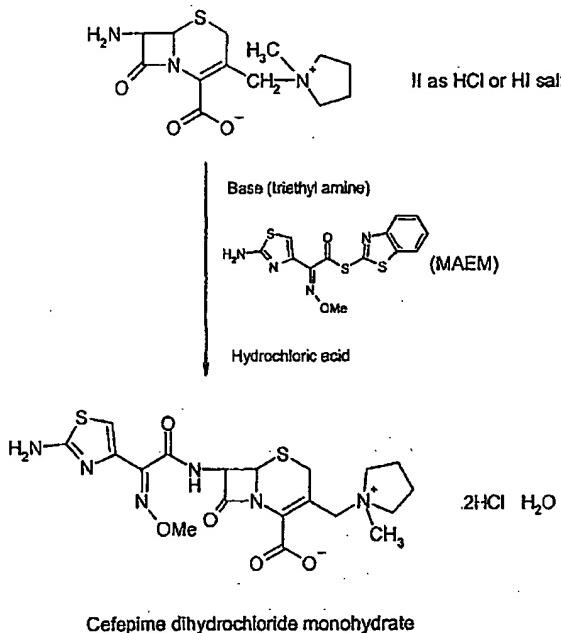
The compound of formula IV may be treated with appropriate HZ or Z⁺ to obtain to the compound of formula III(i) or with appropriate Z to obtain the compound of formula III(ii).

The compounds of formula I, II are readily converted to broad spectrum 10 cephalosporin antibiotics by acylation with the appropriate side-chain acid. Some of the cephalosporin antibiotics that can be prepared include those described in U. S. Patent No. 4,406,899, U. S. Patent No. 4,168309, U. S. Patent No. 4,223,135, U. S. Patent No. 4,336,253, U. S. Patent No. 4,379,787 and J. Organic Chemistry 1988, 53, 983-991. The acylation can be carried out by conventional means using for example acid chloride, 15 mixed acid anhydrides and activated esters. For example the compound of formula II as HCl or HI salt is converted to cefepime dihydrochloride monohydrate by N-acylating with

syn-2-(2-aminothiazol-4-yl)-2-methoxyimino acetyl chloride hydrochloride, syn-2-(2-aminothiazol-4-yl)-2-methoxyimino acetic acid 2-benzothiazolyl thioester (MAEM) or syn-2-(2-aminothiazol-4-yl)-2-methoxyimino acetic acid 1-benzotriazolyl ester and then converting cefepime into cefepime dihydrochloride monohydrate using hydrochloric acid.

5 The preferred method can be shown as in the scheme below:

Scheme



The invention will now be further described by the following examples, which are illustrative rather than limiting.

Example 1

10 7-Aminocephalosporanic acid (7-ACA) (200 gm) is stirred in cyclohexane (1400 ml) for 10 minutes at 25°C and then acetamide (400 mg), imidazole (400 mg) and hexamethyldisilazane (142 gm) are added to the reaction mass at 25°C. The reaction mass is slowly heated to reflux temperature and stirred for 2 hours at the reflux to form a clear solution. The reaction mass is distilled to collect about 100 ml cyclohexane and then the mass is cooled to 5°C to give the reaction mass containing (6R,7R)-3-[(Acetoxy)methyl]-7-(trimethylsilyl)aminoceph-3-em-4-oic acid

15 Trimethylsilyl iodide (246 gm) is slowly added to the mixture of N-methylpyrrolidine (94 gm) and cyclohexane (700 ml) at 5 - 10°C over a period of 30 minutes. Then reaction mass is stirred for 30 minutes at 5 - 10°C. To this mass is added 20 to the reaction mass containing (6R, 7R)-3-[(acetoxy)methyl]-7-

(trimethylsilyl)aminoceph-3-em-4-oic acid over a period of 30 minutes at 5 - 10°C and then trimethylsilyl iodide solution (66 gm in 75 ml cyclohexane) is added at 5 - 10°C in 15 minutes. The mass is heated to 37 - 40°C in 30 minutes and stirred for 35 hours at the same temperature.

- 5 The reaction mass is then cooled to 5°C, isopropyl alcohol (100 ml) is added at 5 - 10°C Concentrated HCl (200 ml) and water (400 ml) are slowly added over a period of 20 minutes at 5 - 10°C. The reaction mass is stirred for 15 minutes. The layers are separated and organic layer is extracted with water (100 ml). Then the combined aqueous layer is cooled to 5 - 10°C, subjected to carbon treatment and filtered on hyflo-bed. The filtrate is cooled to 5°C Isopropyl alcohol (4000 ml) is added to the filtrate over a period of one hour at 5 - 10°C. Then the solid precipitated is filtered, washed with isopropyl alcohol (100 ml) and then dried at 40 - 45°C under vacuum to give 172 gm of [6R-(6 α ,7 β)]-1-[[7-Amino-2-carboxy-8-oxo-5-thia-1-azabicyclo[4.2.0]oct-2-en-3-yl]methyl]-1-methylpyrrolidinium inner salt hydrochloride (HPLC purity 98.77%, 0.08% Δ^2 isomer).

Example 2

[6R-(6 α ,7 β)]-1-[[7-Amino-2-carboxy-8-oxo-5-thia-1-azabicyclo[4.2.0]oct-2-en-3-yl]methyl]-1-methylpyrrolidinium inner salt hydrochloride (25 gm obtained as in example I) is added to a mixture of water (200 ml) and acetone (375 ml) at 5°C and stirred for 10 minutes and syn-2-(2-aminothiazol-4-yl)-2-methoxyimino acetic acid 2-benzothiazolyl thioester (MAEM) (34.10 gm) is added at 5 - 10°C. Triethylamine is slowly added to the reaction mixture at 5 - 10°C to adjust the pH to 7.5 - 7.7 and stirred for 10 minutes at 5 - 10°C. The temperature of the reaction mass is then slowly raised to 20 - 25°C and maintained for 4 hours 30 minutes. Ethyl acetate (250 ml) is added to the reaction mass at 5°C, stirred for 15 minutes and the layers are separated. Then the aqueous layer is extracted with ethyl acetate (125 ml) at 5 - 10°C. The aqueous layer is subjected to carbon treatment and filtered on hyflo-bed. 10 N HCl (60 ml) and acetone (400 ml) are added to the filtrate at 5 - 10°C, seeded with cefepime dihydrochloride monohydrate (0.5 gm) and stirred for 30 minutes at 5 - 10°C. Acetone (850 ml) is added the filtrate for 30 minutes at 5 - 10°C, cooled to 0 - 5°C and maintained for 1 hour. Then the separated solid is filtered, washed with acetone (150 ml) and dried to give 32.8 gm of 7-[α -(2-aminothiazol-4-yl)- α -(z)-methoxyimino acétamido]-3-[(1-methyl-1-pyrrolidinio)methyl]-3-cephem-4-carboxylate dihydrochloride monohydrate (cefepime dihydrochloride monohydrate) (HPLC purity 99.92%, 0.06% Δ^2 isomer).

Example 3

Stage-I:

(6*R*, 7*R*)-Trimethylsilyl 7-(trimethylsilyl)amino-3-acetoxymethylceph-3-em-4-carboxylate:

7-Amino cephalosporanic acid (30 gm) is suspended in cyclohexane (210 ml) at 25°C, then hexamethyldisilazane (27.84 ml), acetamide (60 mg) and imidazole (60 mg) are added at 25°C and the reaction mass is heated to reflux for 3 hours. Then the solution obtained is cooled to 25°C to give the title compound in cyclohexane

5 **Stage-II:**

(6R, 7R)-Trimethylsilyl 7-(trimethylsilyl)amino-3-iodomethylceph-3-em-4-carboxylate:

The solution of (6R, 7R)-Trimethylsilyl 7-(trimethylsilyl)amino-3-acetoxy methylceph-3-em-4-carboxylate in cyclohexane obtained in stage-I is cooled to 0 - 5°C, the solution of trimethylsilyl iodide (48 gm) in cyclohexane (55 ml) is slowly added over a 10 period of 30 minutes and stirred for 1 hour at 0 - 5°C to give the title compound solution in cyclohexane

Stage-III:

(6R, 7R)-Trimethylsilyl 7-(trimethylsilyl)amino-3-(1-methyl-1-pyrrolidinio)methyl ceph-3-em-4-carboxylate iodide:

15 The solution of (6R, 7R)-Trimethylsilyl 7-(trimethylsilyl)amino-3-iodomethylceph-3-em-4-carboxylate in cyclohexane obtained in stage-II is added to a solution of N-methyl pyrrolidine (17.3 ml) in cyclohexane (50 ml) and stirred for 30 minutes at 0 - 5°C. Then the temperature of the reaction mass is raised to 38 - 40°C and stirred for 30 hours to give the title compound solution in cyclohexane.

20 **Stage-IV:**

(6R,7R)-7-amino-3-(1-methyl-1-pyrrolidinio)methylceph-3-em-4-carboxylic acid hydrochloride:

25 The solution of (6R, 7R)-Trimethylsilyl 7-(trimethylsilyl)amino-3-(1-methyl-1-pyrrolidinio)methylceph-3-em-4-carboxylate iodide in cyclohexane as obtained in stage-III is cooled to 0 - 5°C and isopropyl alcohol (15 ml) is slowly added. Then the mixture of concentrated HCl (30 ml) and water (60 ml) is added to the reaction mass at 8 - 10°C and the layers are separated. The aqueous layer is subjected to carbon treatment, filtered and cooled to 8 - 10°C. Then isopropyl alcohol (600 ml) is added slowly to the aqueous layer and the title compound is precipitated. The precipitated solid is filtered, 30 washed with isopropyl alcohol (20 ml) and dried under vacuum at 40°C for 8 hours to 10 hours to give 16 gm of (6R, 7R)-7-amino-3-(1-methyl-1-pyrrolidinio)methylceph-3-em-4-carboxylic acid hydrochloride (0.06% Δ² isomer).

Stage-V:

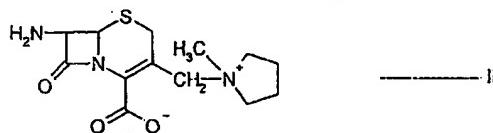
35 Methoxyimino-[2-amino-4-thiazoyl]acetyl chloride hydrochloride (10.9 gm) and (6R,7R)-7-amino-3-(1-methyl-1-pyrrolidinio)methylceph-3-em-4-carboxylic acid hydrochloride (15 gm) are added to a mixture of water (100 ml) and acetone (150 ml) and cooled to 8 - 10°C. The pH of the reaction mass is adjusted to 7.2 - 7.5 with

triethylamine and then stirred for 4 hours at 10⁰C. Ethyl acetate (150 ml) is added to the reaction mass, stirred for 30 minutes and separated the layers. The aqueous layer is then subjected to carbon treatment, stirred for 30 minutes and filtered. The mixture of acetone (36 ml) and concentrated HCl (36 ml) is added to the filtrate at 5⁰C. Acetone (700 ml) is added and cooled to 0 ~ 5⁰C. Then the separated solid is filtered, washed with acetone (50 ml) and dried under vacuum at 40⁰C for 10 hours to give 18 gm of 7-[α -(2-aminothiazol-4-yl)- α -(Z)-methoxyiminoacetamido]-3-[(1-methyl-1-pyrrolidinio)methyl]-3-cephem-4-carboxylate dihydrochloride monohydrate (cefepime dihydrochloride mono hydrate) (HPLC purity 99.82%, 0.05% Δ^2 isomer).

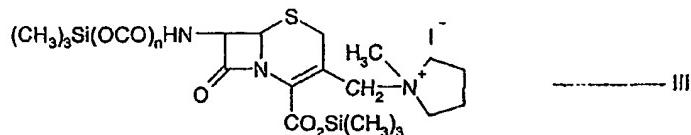
10

We claim:

1. A process for the preparation of the compound of formula II:



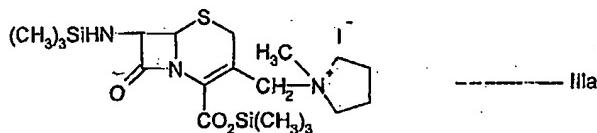
or a salt thereof which is substantially free of the Δ^2 isomer, which comprises treating
5 the compound of formula III:



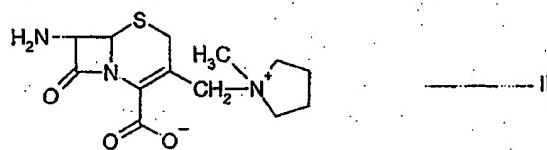
wherein n = 0 or 1,

in cyclohexane with a C₁ - C₄-alkanol or water to remove silyl protecting groups,
optionally converting to the salt of the compound of formula II.

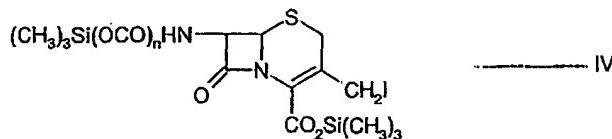
- 10 2. The process according to claim 1, wherein the salt is hydrochloride or hydroiodide salt.
3. The process according to claim 1, wherein the compound of the formula III used is the compound IIIa;



- 15 4. The process according to claims 1 and 3, wherein the C₁ - C₄- alkanol is selected from the group consisting of isopropyl alcohol, methanol and ethanol.
5. The process according to claim 4, wherein the C₁ - C₄- alkanol is isopropyl alcohol.
6. A process for the preparation of the compound of formula II:

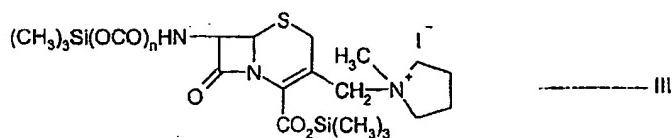


- 20 20 or a salt thereof which is substantially free of the Δ^2 isomer, comprising the steps of:
a) reacting the compound of formula IV:



wherein n = 0 or 1,

in cyclohexane with N-methylpyrrolidine to produce the compound of formula III:



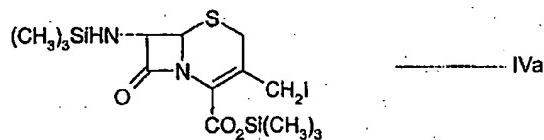
5 wherein n = 0 or 1,

and

(b) treating the compound of formula III in cyclohexane with a C₁ - C₄-alkanol or water to remove silyl protecting groups, optionally converting to the salt of the compound of formula II.

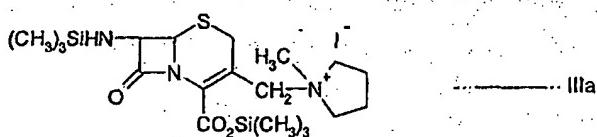
10 7 The process according to claim 6, wherein the conversion into the salt in step (b) is carried out by treating the compound of formula II with hydrochloric acid or hydroiodic acid.

8. The process according to claims 6 and 7, wherein the compound of formula IV used is the compound IVa:

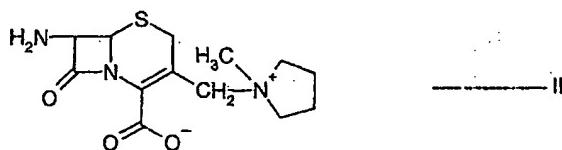


15

to obtain the compound formula IIIa:

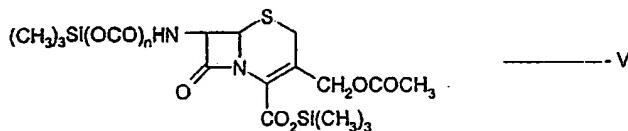


9. A process for the preparation of the compound of formula II:



or a salt thereof which is substantially free of the Δ^2 isomer, comprising the steps of:

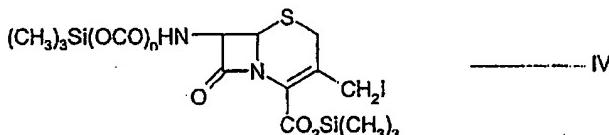
- a) reacting the compound of formula V:



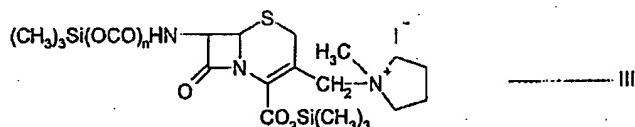
wherein n = 0 or 1,

- 5 in cyclohexane with at least one equivalent of trimethylsilyl iodide per equivalent of compound of formula V to produce the compound of formula IV:

wherein n = 0 or 1,



- b) reacting the compound of formula IV in cyclohexane with N-methylpyrrolidine to produce the compound of formula III:



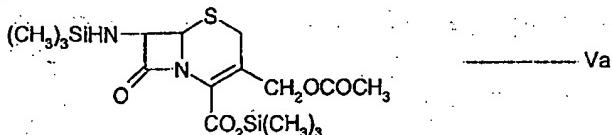
10 wherein n = 0 or 1,

and

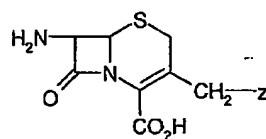
- (c) treating the compound of formula III in cyclohexane with a C₁ - C₄-alkanol or water to remove silyl protecting groups, optionally converting to the salt of the compound of formula II.

15 10. The process according to claim 9, wherein the salt is hydrochloride or hydroiodide salt.

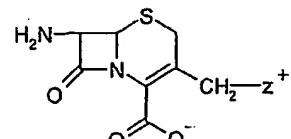
11. The process according to claim 9, wherein the compound of the formula V used is the compound Va;



20 12. A process for the preparation of the compound of formula I(i) or I(ii):



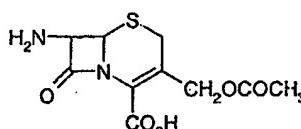
I(i)



I(ii)

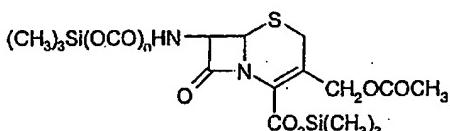
or a salt thereof which is substantially free of the Δ^2 isomer, comprising the steps of:

- a) treating the compound of formula VI:



VI

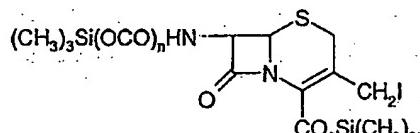
- 5 in cyclohexane with at least one equivalent of hexamethyldisilazane per equivalent of compound of formula VI and catalytic amount of trimethylsilyl iodide to produce the compound of formula V:



V

wherein $n = 0$ or 1 ,

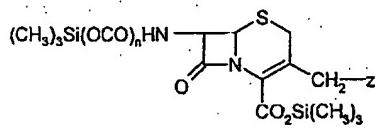
- b) treating the compound of formula V in cyclohexane with at least one equivalent of trimethylsilyl iodide per equivalent of compound of formula V to produce the compound of formula IV:



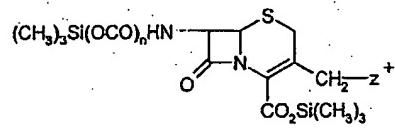
IV

wherein $n = 0$ or 1 ,

- c) reacting the compound of formula IV in cyclohexane with Z^- or HZ to produce the compound of formula III(i) or with Z to produce the compound of formula III(ii):



III(i)

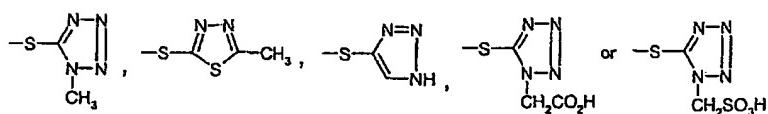


III(ii)

15

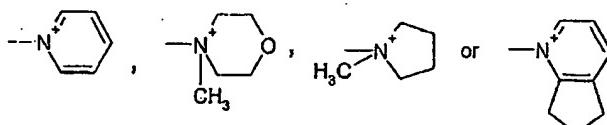
wherein $n = 0$ or 1 ,

乙



and

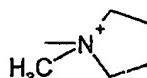
Z^+ is



- (d) treating the compound of formula III(i) or III(ii) in cyclohexane with a C₁ - C₄-alkanol or water to remove silyl protecting groups, optionally converting to the salt of the compound of formula II.

5

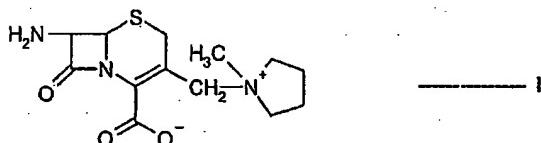
13. The process according to claim 12; wherein the compound produced in step (c) is III(ii) wherein Z⁺ is



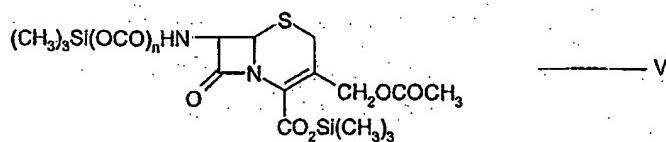
and $n = 0$.

14. The process according to claim 12, wherein the salt is hydrochloride or hydroiodide
10 salt.

15. A process for the preparation of the compound of formula II:



or a salt thereof which is substantially free of the Δ^2 isomer, which comprises treating a solution of the compound of formula V:



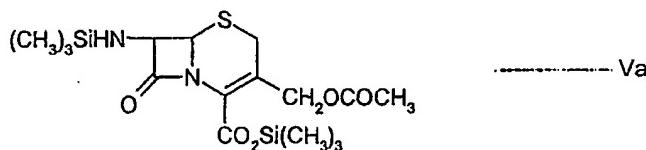
15

wherein $n = 0$ or 1 ,

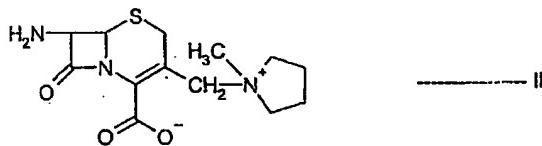
in cyclohexane with at least one equivalent of N-methylpyrrolidine then with at least one equivalent of trimethylsilyl iodide per equivalent of compound of formula V,

followed by treatment with a C₁ - C₄ - alkanol or water to remove silyl protecting groups, optionally converting to the salt of the compound of formula II.

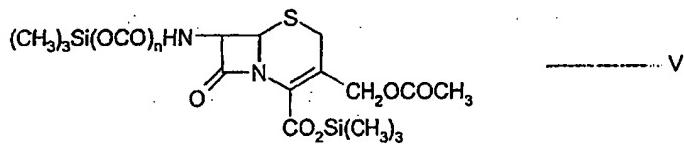
16. The process according to claim 15, wherein the salt is hydrochloride or hydroiodide salt.
- 5 17. The process according to claim 15, wherein the compound of the formula V used is the compound Va;



18. A process for the preparation of the compound of formula II:



- 10 or a salt thereof which is substantially free of the Δ² isomer, which comprises treating a solution of the compound of formula V:

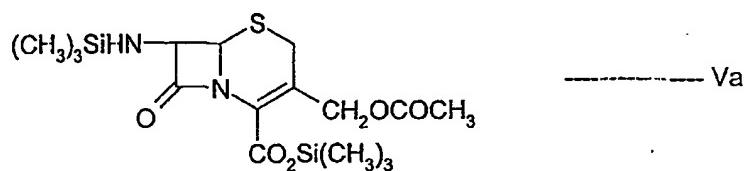


wherein n = 0 or 1,

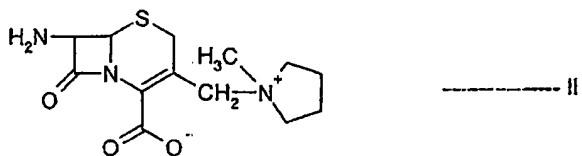
in cyclohexane with the compound of formula VII:



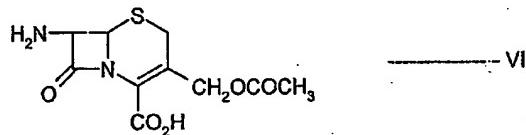
- 15 in cyclohexane, followed by treatment with a C₁ - C₄ - alkanol or water to remove silyl protecting groups, optionally converting to the salt of the compound of formula II.
19. The process according to claim 18, wherein the salt is hydrochloride or hydroiodide salt.
20. The process according to claim 18, wherein the compound of the formula V used is the compound Va;



21. A process for the preparation of the compound of formula II:



or a salt thereof which is substantially free of the Δ^2 isomer, which
5 comprises treating a solution of the compound of formula VI:



in cyclohexane with at least one equivalent of hexamethyldisilazane per equivalent
of compound VI and then with the compound of formula VII:



- 10 in cyclohexane, followed by treatment with a C₁ - C₄-alkanol or water to remove silyl protecting groups, optionally converting to the salt of the compound of formula II.
- 22 The process according to claim 21, wherein the salt is hydrochloride or hydroiodide salt.
- 23 The process according to claim 21, wherein the reaction with the compound VII is carried out in the presence of trimethylsilyl iodide
- 15 24. The process according to claim 21, wherein the reaction with hexamethyldisilazane is carried out in the presence of the catalytic amounts of imidazole and acetamide.
- 25. The process according to claim 21, wherein the reaction with hexamethyldisilazane is carried out in the presence of the catalytic amount of trimethylsilyliodide.

INTERNATIONAL SEARCH REPORT

International application No.
PCT/IN 2004/000209

A. CLASSIFICATION OF SUBJECT MATTER IPC⁷: C 07 D 501/18 According to International Patent Classification (IPC) or to both national classification and IPC		
B. FIELDS SEARCHED Minimum documentation searched (classification system followed by classification symbols) IPC⁷: C 07 D		
Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched		
Electronic data base consulted during the international search (name of data base and, where practicable, search terms used) REGISTRY, CAPLUS, WPI, PAJ, EPODOC		
C. DOCUMENTS CONSIDERED TO BE RELEVANT		
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	EP 581 220 A2 (BRISTOL-MYERS SQUIBB COMPANY), 2 February 1994 (02.02.1994) <i>the whole document.</i> --	1-25
A	D. G. WALKER et al., "Use of Bistrimethylsilylated Intermediates in the Preparation of Semisynthetic 7-Amino-3-substituted-cephems. Expedient Syntheses of a New 3-[(1-Methyl-1-pyrrolidinio)methyl]cephalosporin", J. Org. Chem., 1988, 53, pages 983-991 <i>the whole document.</i> --	1-25
<input checked="" type="checkbox"/> Further documents are listed in the continuation of Box C. <input checked="" type="checkbox"/> See patent family annex.		
* Special categories of cited documents: "A" document defining the general state of the art which is not considered to be of particular relevance "E" earlier application or patent but published on or after the international filing date "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified) "O" document referring to an oral disclosure, use, exhibition or other means "P" document published prior to the international filing date but later than the priority date claimed		
Date of the actual completion of the international search 4 October 2004 (04.10.2004)	Date of mailing of the international search report 14 October 2004 (14.10.2004)	
Name and mailing address of the ISA/ AT Austrian Patent Office Dresdner Straße 87, A-1200 Vienna Facsimile No +43 / 1 / 534 24 / 535	Authorized officer WENIGER S. Telephone No. +43 / 1 / 534 24 / 341	

INTERNATIONAL SEARCH REPORT

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C (Continuation). DOCUMENTS CONSIDERED TO BE RELEVANT		
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	WO 1987/001116 A1 (BRISTOL-MYERS COMPANY), 26 February 1987 (26.02.1987) <i>the whole document, esp. claims.</i>	1-25
A	-- EP 237 735 A2 (FUJISAWA PHARMACEUTICAL CO., LTD.), 23 September 1987 (23.09.1987) <i>the whole document.</i>	1-25
A	-- WO 1986/003204 A1 (BIOCHEMIE GESELLSCHAFT M.B.H.), 5 June 1986 (05.06.1986) <i>the whole document.</i>	1-25

INTERNATIONAL SEARCH REPORT

Information on patent family members

International application No.
PCT/IN 2004/000209

Patent document cited in search report	Publication date	Patent family member(s)			Publication date
A		none			
EP A 237735		CN	A 86105998	1987-03-11	
		HU	A 43077	1987-09-28	
		NO	A 862910	1987-02-06	
		ES	A 2000822	1988-03-16	
		PT	A 83141	1986-09-01	
		AT	T 103607T	1994-04-15	
EP A 581220		IL	A 106388	1999-10-28	
		DK	T 581220T	1999-12-20	
		GR	T 3032045T	2000-03-31	
		DE	T 69326400T	2000-05-18	
		ES	T 2137211T	1999-12-16	
		DE	D 69326400D	1999-10-21	
WO A 19860032		none			
04					
WO A 19870011		none			
16					

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